SYNTHESIS OF A NEW SULPHUR-CONTAINING CARBAPROSTACYCLIN ANALOGUE P.G. Baraldi a , A. Barco b , S. Benetti b , C.A. Gandolfi c , G.P. Pollini a and

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ABSTRACT: The synthesis of tetrahydro- \underline{cis} -cyclopenta [c]thiophen-5-one ethylene ketal $\underline{5}$ and its transformation into 11-thia-carbacyclin analogues are described.

Most of the biological and pharmacological activities of prostacyclin (PGI $_2$) $\underline{1}$ are still retained when its characteristic bicyclo $\underline{[}3.3.0\underline{]}$ octane framework was replaced by a perhydropentalene nucleus $\underline{[}1$. Other changes made in the same skeleton (for instance 4-oxabicyclo- $\underline{[}2$ and 2-thiabicyclo $\underline{[}3.3.0\underline{]}$ octane $\underline{[}3$ analogues) did not appear fortunate enough to yield close mimics of the natural substance. The presence of a C-11 hydroxy group seems to be an essential requisite for a biological response $\underline{[}4$. A recent report $\underline{[}5$ showed that in one case the lack of the C-11 hydroxy group in carbaanalogs seems to induce $\underline{[}1$ vivo a selective antagonism against blood pressure lowering activity of the PGI $_2$ putting out in evidence the dramatic role of this C-11 appendix in recognizing the receptor site.

To furtherly probe this matter, we planned the synthesis of the novel stable analogue $\underline{3}$, carrying a sulphur atom in place of the hydroxymethine group in C-11 position.

We chose the sulphur analogue $\underline{5}$ of the monoketal of $\underline{\text{cis}}$ -bicyclo $\begin{bmatrix} 3.3.0 \end{bmatrix}$ octane-3,7-dione $\underline{4}$, a classical intermediate in carbacyclin synthesis 1 , as a bifunctional starting material.

This synthon offers the opportunity to introduce the required side-chains taking advantage of the protected carbonyl group and of the acidity of the hydrogen atoms adjacent to the sulphoxide moiety.

The preparation of $\underline{5}$ was achieved as outlined in the scheme starting from the known tetrahydro- $\underline{\mathrm{cis}}$ -cyclopenta[c] furan-5-one, $\underline{6}^6$, which was transformed by standard steps into the bicyclic sulphide $\underline{9}^*$.

Oxidation of $\underline{9}$ with sodium metaperiodate proceeded cleanly to give almost exclusively $\underline{5}$, where the S-O bond is \underline{cis} to the junction hydrogens as can be deduced from the NMR studies on related compounds⁷.

The introduction of the ω -chain was performed by reaction of the carbanion generated from $\underline{5}$ with LDA in THF with the protected aldehyde $(\underline{10})^8$ producing an 80% yield of an unseparable mixture of aldols $\underline{11}$.

Diborane promoted deoxygenation followed by deketalization (80% aqueous acetic acid) led to the β -hydroxyketone 12, which was dehydrated with 20% sulfuric acid to afford the $\alpha\beta$ -trans-enone 13 (75% overall yield from 5).

$$\frac{1) B_{2} H_{6}}{C_{5} H_{11} - C - C H_{2} - C HO} \\
\underline{10} \\
\underline{10} \\
\underline{11} \\
\underline{10} \\
\underline{11} \\
\underline{11$$

The two carbonyl functions of $\underline{13}$ were chemoselectively differentiated by ketal exchange with 2-methyl-2-ethyl-1,3-dioxolane (MED) at room temperature to produce quantitatively $\underline{14}$, which was reduced with NaBH $_4$ at 0°C to give an unseparable mixture of the epimeric alcohols $\underline{15a}$,b. Removal of the ketal protective group by action of dilute sulfuric acid allowed the separation of the epimers $\underline{16a}$ and $\underline{16b}$ by column chromatography on silica gel, as oils with almost identical spectroscopic properties. Condensation of the single C-15 epimers with (4-carboxybutylene)triphenyl-phosphorane in DMSO in the presence of K tert-butoxide for 3h at 30°C gave rise to $\underline{3}$ and its 15(R)-epimer, which were obtained as homogeneous oils after chromatography on silica gel (eluent Et $_2$ 0-Petroleum ether 3:1), in 30% and 16% yield respectively, as unseparable E/Z mixture.

The two new 11-deoxy-11-thia carbaprostacyclin analogues are characterized by a very weak antiaggregatory activity and are not able to antagonize prostacyclin in vivo.

References and Notes

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- ** The preferential formation of $\underline{13}$ may be accounted for by the chair-like six center transition state \underline{A} in the aldolization process which favours the delivering of the chain \underline{cis} to the S 0 bond and \underline{trans} to the cyclopentane ring.

H S C₅H₁₁

NMR (CDCI₂) and IR data are given for:

<u>5</u>: $\delta = 1.5 - 2.3$ (m, 4H); 2.5 - 2.8 (m, 2H); 2.9 - 3.5 (m, 4H); 3.9 (s, 4H). m.p. 70 - 71°C.

<u>13</u>: δ =0.9 (t, 3H, J=4Hz); 1.1-1.9 (m, 8H); 2.1-3.5 (m, 8H); 3.7 (dd, 1H, J=8Hz), 6.1 (d, 1H, J=15Hz); 6.6 (dd, 1H, J=15, J=8Hz). IR (film): 1740, 1720, 1620, 990 cm⁻¹.

 $\underline{3}:\delta=0.9$ (t, 3H, J=4Hz); 1-3.3 (m, 23H); 3.4-3.7 (m, 1H); 3.9-4.2 (m, 1H); 4.5-5.3 (m, 3H); 5.4-5.6 (m, 2H). I.R. (film): 3300, 1710 cm⁻¹.

*** We thank Dr. R. Ceserani and M. Bergamaschi for these preliminary biological results.

(Received in UK 30 June 1983)